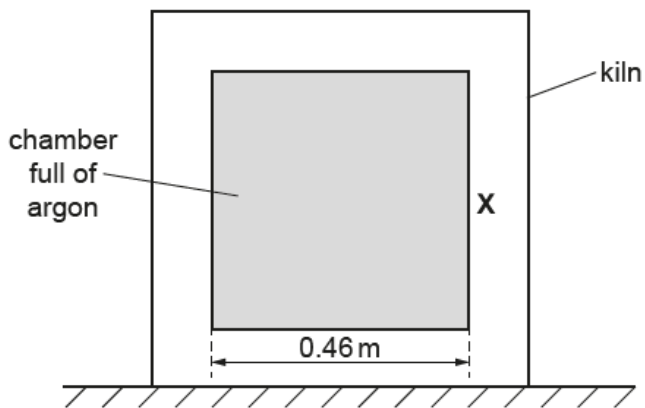


Density and Pressure – 2021/20 GCE Physics A Component 01**1. Nov/2020/Paper_H556_01/No.21(a)**

A kiln used to harden ceramics is shown below.



The internal chamber is a cube. Each side of this cube has length 0.46 m. The chamber is sealed and full of argon. Argon behaves as an ideal gas.

- (a) The kiln is initially at 20 °C.
The argon in the kiln has an initial pressure of 100 kPa.
- (i) Calculate the amount of argon n in the chamber in moles.

$$n = \dots\dots\dots \text{ mol [2]}$$

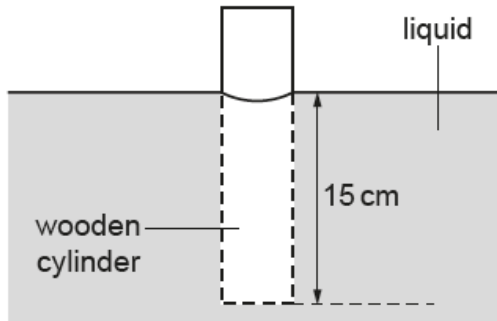
- (ii) The temperature of the kiln is increased from 20 °C to 1300 °C.

Calculate the pressure in kPa at 1300 °C.

$$\text{pressure} = \dots\dots\dots \text{ kPa [2]}$$

2. Nov/2020/Paper_H556_01/No.22(a)

A long wooden cylinder is placed into a liquid and it floats as shown.



The length of the cylinder below the liquid level is 15 cm.

(a) (i) State **Archimedes' principle**.

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 [1]

(ii) The pressure exerted by the liquid alone on the bottom of the cylinder is $1.9 \times 10^3 \text{ Pa}$.

Calculate the density ρ of the liquid.

$\rho = \dots\dots\dots \text{ kg m}^{-3}$ [2]