<u>Interpreting and interacting with earth systems – 2021/20 GCSE Gateway Chemistry Combined Science A</u>

1.

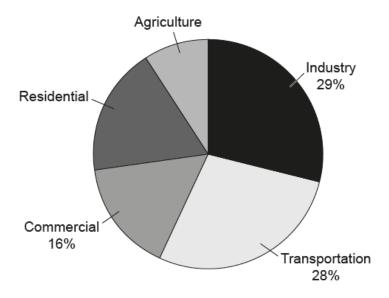
2.

3.

	v/2021/Paper_J250/04/No.1 e burning of fossil fuels can produce gases that cause acid rain.	
Wh	nich gas causes acid rain?	
Α	Carbon monoxide	
В	Nitrogen	
С	Oxygen	
D	Sulfur dioxide	
Yo	ur answer	[1]
	/2021/Paper_J250/04/No.4 Earth's early atmosphere contained little or no oxygen.	
Whi	ich process produced the oxygen found in the Earth's atmosphere today?	
Α	Burning fossil fuels	
В	Photosynthesis by plants and algae	
С	Respiration by bacteria	
D	Volcanic activity	
Mai	/2021/Paper_J250/04/No.6 ny scientists believe that increased levels of methane in the atmosphere are contributing bal warming.	j to
Wh	at causes increased levels of methane in the atmosphere?	
Α	Cutting down trees	
В	Exhaust gases from car engines	
С	Incomplete combustion of fossil fuels	
D	Waste gases from cows and landfill sites	
Υοι	ur answer	[1]

4. Nov/2021/Paper_J250/04/No.10

The diagram shows the percentage of greenhouse gases made from different sources.



The percentage of greenhouse gases produced from Residential is **twice** that produced from Agriculture.

What is the percentage of greenhouse gases produced from Residential?

- A 9%
- **B** 18%
- C 27%
- **D** 36%

Your answer [1]

5. Nov/2021/Paper_J250/04/No.16

Fig. 16.1 shows how the atmospheric carbon dioxide concentration changed between 1970 and 2010.

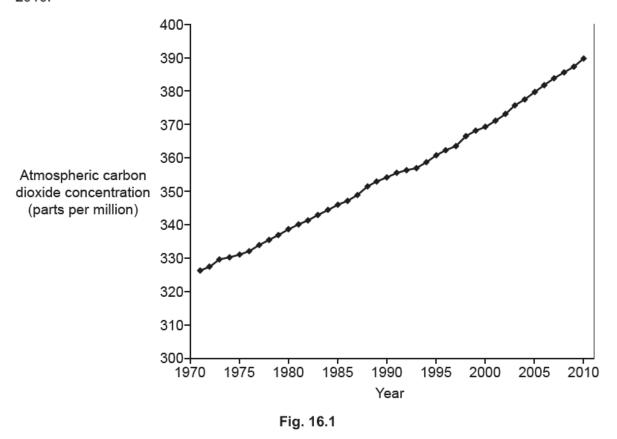
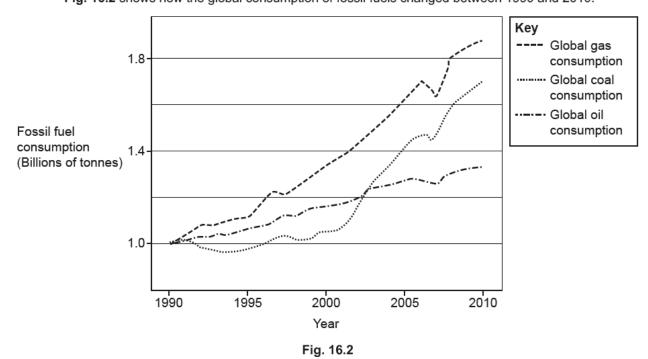


Fig. 16.2 shows how the global consumption of fossil fuels changed between 1990 and 2010.



Describe and explain if the graphs in Fig. 16.1 and Fig. 16.2 support this statement.

A student thinks that the increase in atmospheric carbon dioxide is because of the increased consumption of fossil fuels.

Jse information from Fig. 16.1 and Fig. 16.2 in your answer.				

6. Nov/2020/Paper_J250/04/No.11

4.5 billion years ago the Earth's atmosphere was different from the atmosphere today.

The table shows the gases found in the Earth's atmosphere as it is today.

(a) Calculate the percentage of argon in the atmosphere.

Gas	Nitrogen	Oxygen	Argon	Carbon dioxide	Other gases
Percentage in atmosphere (%)	78	21		0.04	0.06

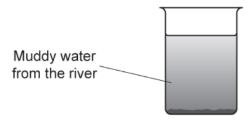
				Per	centage =			% [1]
(b)		k at the list rs ago.	of gases that i	may have t	been present i	n the Earth's	atmosphere	4.5 billion
	amr	nonia	carbon dioxid	de ca	ırbon monoxid	de		
	hyd	rogen	methane	water va	ipour			
			owing questions					
	Eac	h word may	be used once,	more than	once, or not at	: all.		
	(i)	Which gas	condensed to f	orm the oce	eans?			
	(ii)		was turned into	oxygen by		gae?		
								[1]
	(iii)	Which gas	was turned into	nitrogen b	y bacteria?			
								[1]

7. Nov/2020/Paper_J250/04/No.14

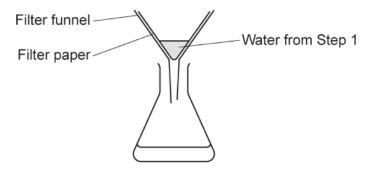
Drinking water which comes from rivers needs to be made safe to drink.

- (a) What is the name of water that is safe to drink?
 -[1]
- (b) A student wants to produce water that is safe to drink from muddy water from a river.

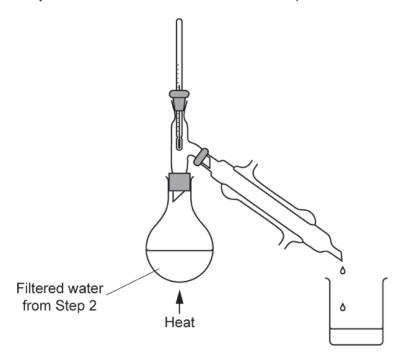
Step 1: He leaves the water from the river in a beaker for 24 hours.



Step 2: He filters the top half of the water in the beaker from Step 1.



Step 3: He distils the filtered water from Step 2.



Explain how e	ach step purifies the water.	
Step 1		
Step 2		
Step 3		
		[3]
	be turned into drinking water by desalination. T smosis and simple distillation.	here are two main methods for
		Approximate cost of drinking water (£ per m³)
this, reverse o	smosis and simple distillation.	Approximate cost of
Name of method Reverse	Description of method Seawater is pumped through 'ultrafilters',	Approximate cost of drinking water (£ per m³)
Name of method Reverse osmosis Simple distillation	Description of method Seawater is pumped through 'ultrafilters', which trap salt and produce drinking water. Seawater is heated to separate pure water	Approximate cost of drinking water (£ per m³) 30 80
Name of method Reverse osmosis Simple distillation	Description of method Seawater is pumped through 'ultrafilters', which trap salt and produce drinking water. Seawater is heated to separate pure water from salt which is left behind.	Approximate cost of drinking water (£ per m³) 30 80 everse osmosis.
Name of method Reverse osmosis Simple distillation (i) Suggest v	Description of method Seawater is pumped through 'ultrafilters', which trap salt and produce drinking water. Seawater is heated to separate pure water from salt which is left behind. why simple distillation is more expensive than results.	Approximate cost of drinking water (£ per m³) 30 80 everse osmosis.

(c)

8. Nov/2020/Paper_J250/04/No.15

Shopping bags can be made from different types of plastics.

The table shows information about three different plastic bags.

	Weight Volume		For every 100 bags made:		
Plastic bag	Weight (g)	(cm ³)	Energy used (kJ)	Waste produced (g)	
Α	8	19000	2.0 × 10 ³	42	
В	35	21 000	1.7 × 10 ⁴	17	
С	116	20 000		585	

(a)	Plastic bag	C uses	31600 J	of energy	for every	one bag made.	
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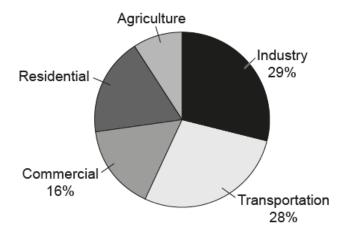
Calculate the energy used (in kJ) for every 100 bags of Plastic bag C made.

Write your answer in standard form in the table.

		[3]
(b)	One student thinks that Plastic bag A is the best bag to use.	
	Another student thinks Plastic bag B is the best bag to use.	
	Explain why both students could be correct.	
	Use information from the table in your answer.	

9. Nov/2021/Paper_J250/10/No.2

The diagram shows the percentage of greenhouse gases made from different sources.



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What is the percentage of greenhouse gases produced from Residential?

- A 9%
- **B** 18%
- C 27%
- **D** 36%

Your answer	[1]
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10. Nov/2021/Paper_J250/10/No.5

Which row correctly shows a gas produced in a car engine and the problem it can cause?

	Gas produced by a car engine	Problem caused by the gas	
A Water vapor		Toxic to humans	
В	Carbon monoxide	Lung disease	
С	Methane	Breathing difficulties	
D	Nitrogen dioxide	Acid rain	

Your answer	[1]
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11. Nov/2021/Paper_J250/10/No.6

The table shows the composition of the atmosphere of four different planets.

Planet	Composition of the planet's atmosphere					
Α	Carbon dioxide Nitrogen 96% 3%		Other gases 1%			
В	Hydrogen 80%	Helium 19%	Methane 1%			
С	Nitrogen 97%	Methane 2.5%	Carbon monoxide 0.5%			
D	Nitrogen 78%	Oxygen 21%	Other gases 1%			

	Wh	nich planet, A, B, C or D, has an atmosphere closest to the Earth's early atmosphere?			
	You	ur answer	[1]		
12. Nov/2020/Paper_J250/10/No.5 The composition of the Earth's atmosphere has changed over the last 4.5 billion years.					
	Whi	ch gas has increased by the largest amount?			
	Α	Ammonia			
	В	Nitrogen			
	С	Oxygen			
	D	Water vapour			
	You	r answer	[1]		

13. Nov/2020/Paper_J250/10/No.9

Your answer

The Earth's early atmosphere is thought to have contained large amounts of carbon dioxide.

Which process does **not** explain how the amount of carbon dioxide in the atmosphere was decreased?

Α	It became 'locked up' in fossil fuels.
В	It dissolved in the oceans.
С	It reacted to produce metal oxides found in metal ores.
D	It was used in photosynthesis by green plants.

[1]

14. Nov/2020/Paper_J250/10/No.15

This question is about the Earth's atmosphere and the greenhouse effect.

(a)	Scientists think that without the greenhouse effect the average temperature of the Earth's surface would be about -18°C .
	Explain how the greenhouse effect keeps the average temperature above −18 °C.
	[2]
(b)	Many scientists think that human activity is enhancing the greenhouse effect and leading to global warming.
	Explain this statement.
	[2]
(c)	Student ${\bf A}$ thinks that the increased use of electric cars would help reduce the greenhouse effect.
	Student ${\bf B}$ thinks that the increased use of electric cars would ${\bf not}$ help reduce the greenhouse effect.
	Explain why each student may be correct.
	[2]

(d) The waste gases from vehicle engines contain oxides of nitrogen.

(i)	Explain how the oxides of nitrogen are formed in an engine.
	[2]
(ii)	Most vehicles are fitted with catalytic converters. They reduce the amount of the oxides of nitrogen released into the atmosphere.
	Describe why this is important.
	[1]